

Solid Phase Peptide Synthesis

Current Pure and Applied Chemistry

Linda Haugaard-Kedström



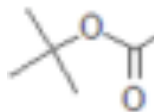
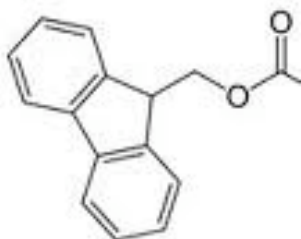
Outline

- History
 - Introduction to solid phase peptide synthesis
 - Resins and linkers
 - Protecting groups
 - Activation
 - Deprotection
 - Cleavage
 - Mechanisms
 - Boc and Fmoc
 - Purification and characterisation
-
- Leu-Enkephalin



Bruce Merrifield

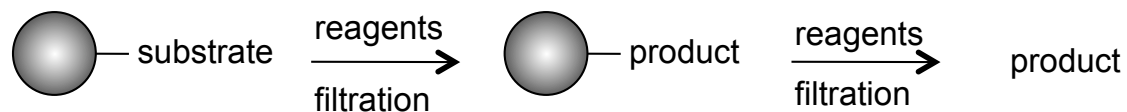
- Pioneered the field
- Published the synthesis of a tetrapeptide in 1963 (article can be found on Moodle)
- Nobel prize in chemistry 1984
“for his development of methodology for chemical synthesis on a solid matrix”
- Boc or t-Boc (tert-butyloxycarbonyl)
- Fmoc (9-fluorenylmethyloxycarbonyl)



Introduction to SPPS

What is SPPS?

- Organic reactions carried out on substrates that are covalently attached to a polymeric resin



Traditional solution phase synthesis

- Important with good protecting groups

Advantages with SPPS

Why is solid-phase (often) nicer than “traditional synthesis”?

- Synthetic intermediates don't have to be isolated
- Excess reagents are used —→ drive reactions to completion
- Reagents are simply washed away each step
- Overall quicker
- Can be automated with robots



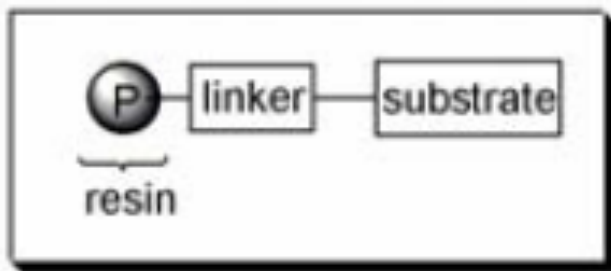
General mechanism SPPS

On white board

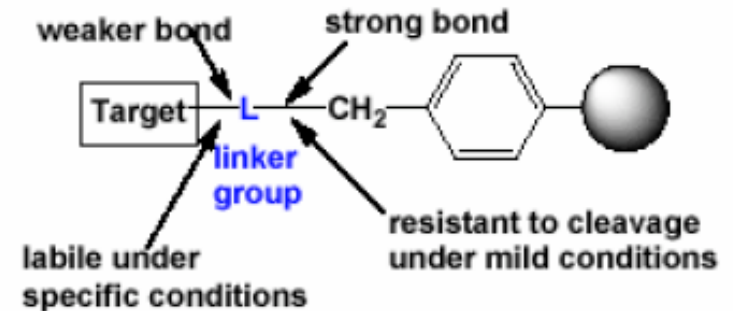


Resins and linkers

- Synthesis occurs on multiple surface of the bead and inside the bead



Resins and linkers

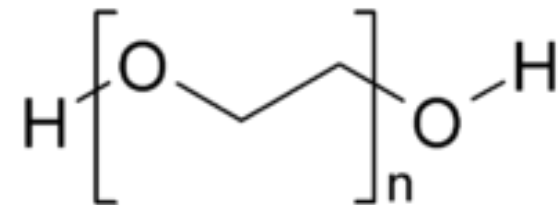
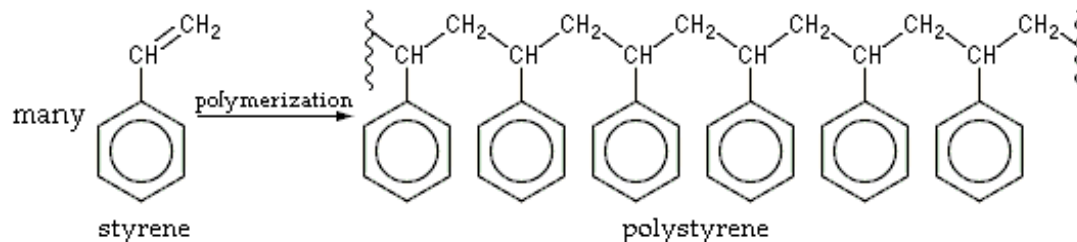


What is a resin?

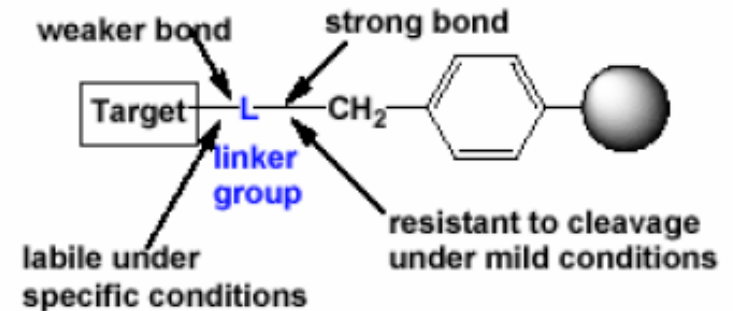
- Polymeric backbone that substrate is anchored to
- Different resins have different swelling properties

Examples:

- Polystyrene (PS): Swells in non-polar solvents
- Polyethylene glycol (PEG): Swells in polar and non-polar solvents
- Many others exist



Resins and linkers



What is a resin?

- Polymeric backbone that substrate is anchored to
- Different resins have different swelling properties

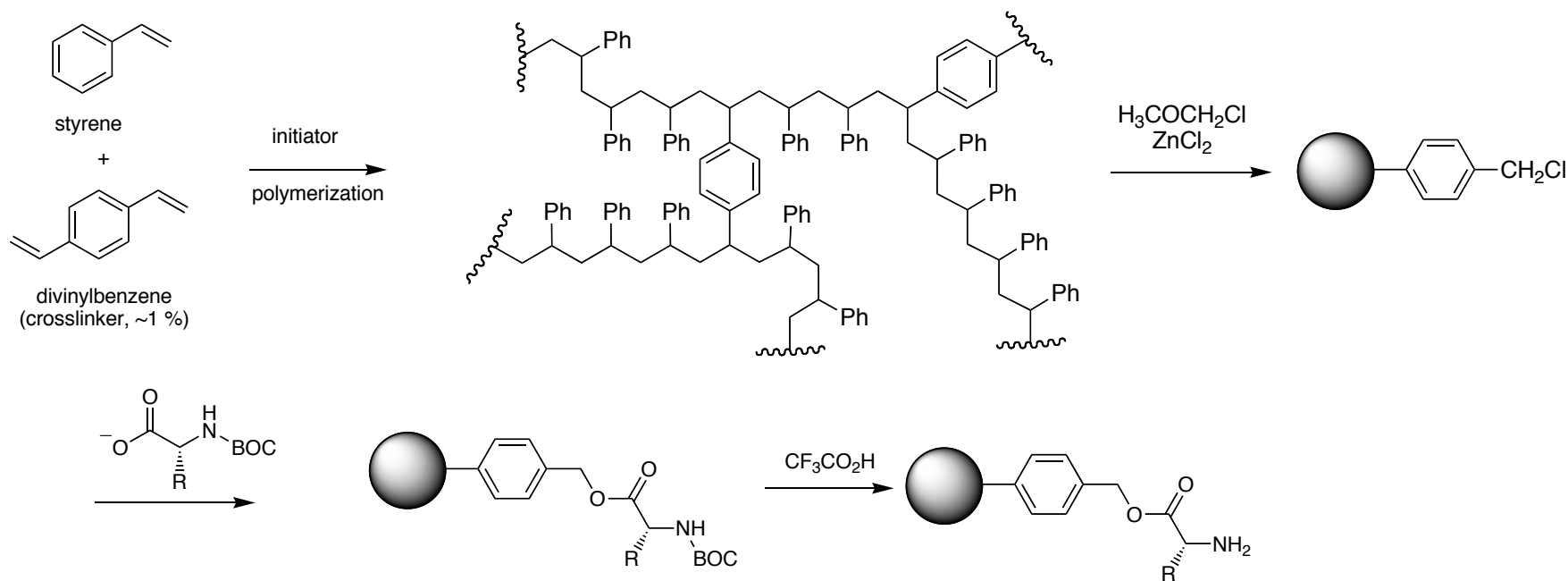
Examples:

- Polystyrene (PS): Swells in non-polar solvents
- Polyethylene glycol (PEG): Swells in polar and non-polar solvents
- Many others exist

What is a linker?

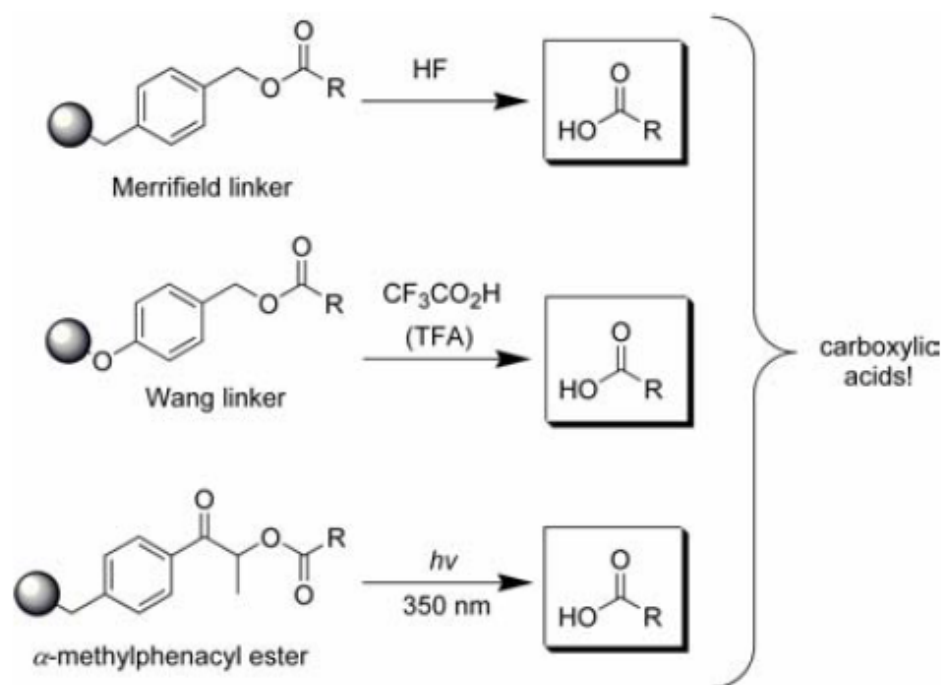
- An intermediate structure between resin and substrate
- Different linkers can be used to unmask different functional groups on the substrate

The solid support (Merrifield resin): polystyrene polymer



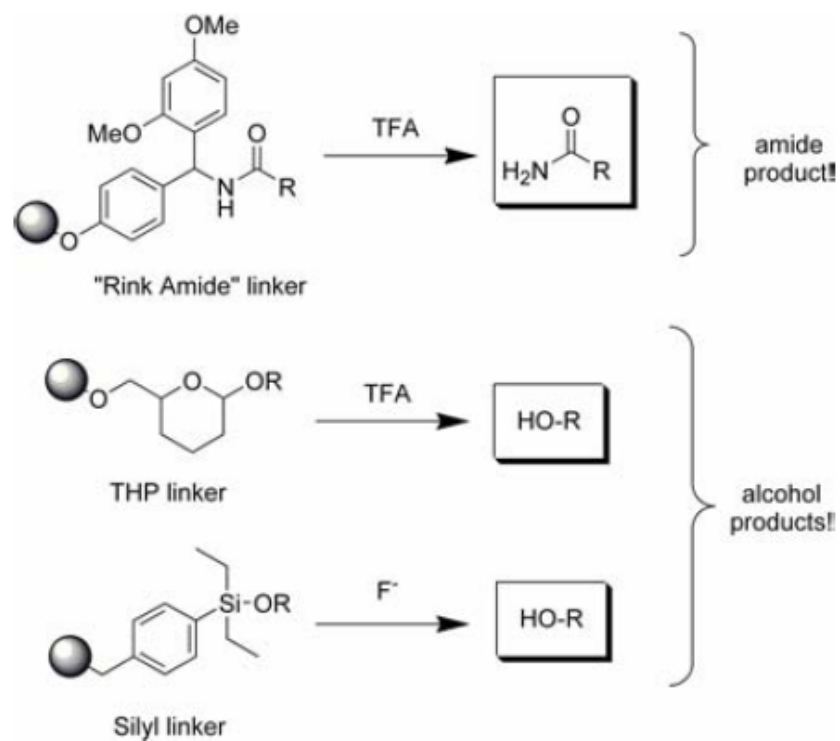
Linkers

Examples of linkers commonly used in solid phase peptide synthesis:

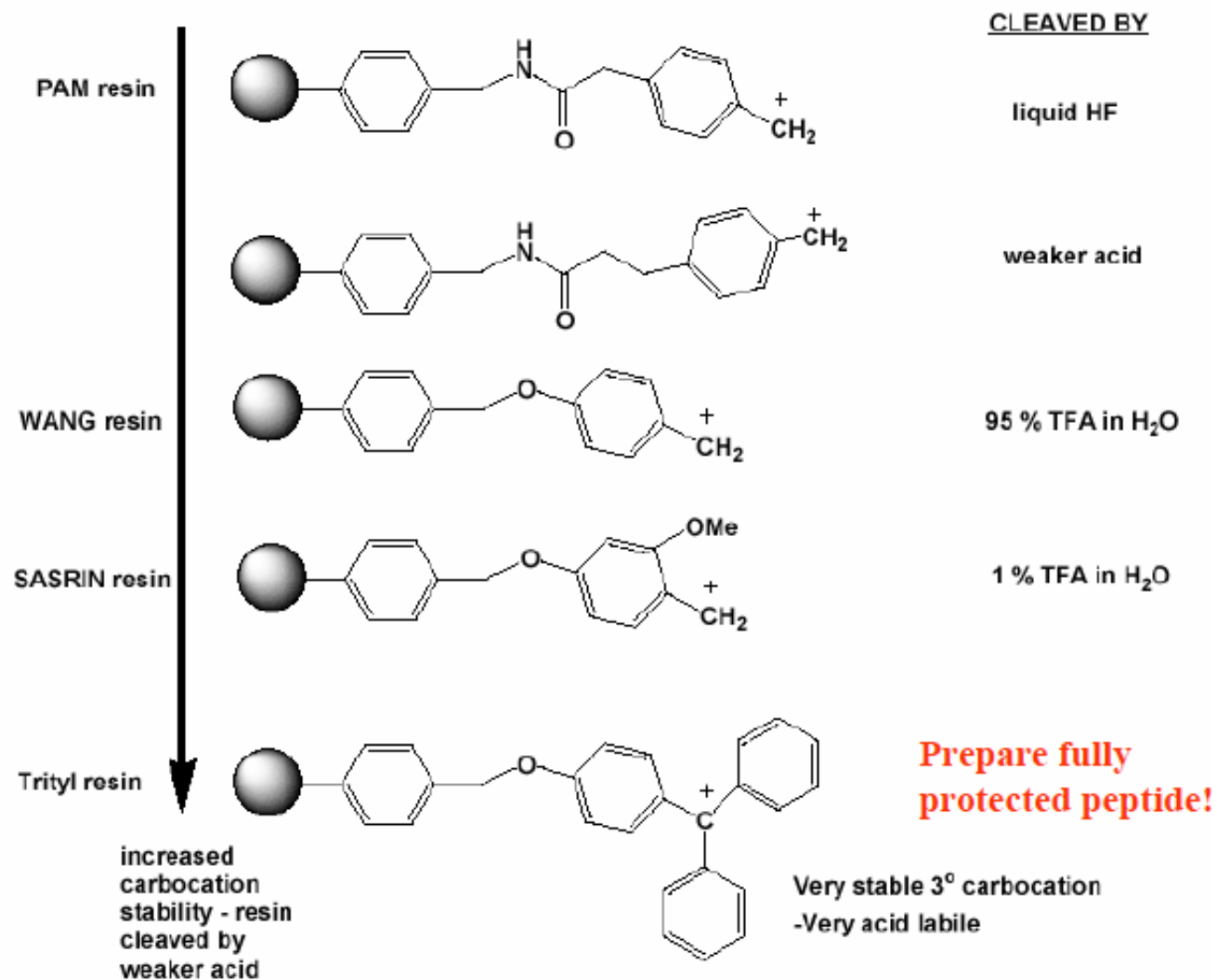


Linkers

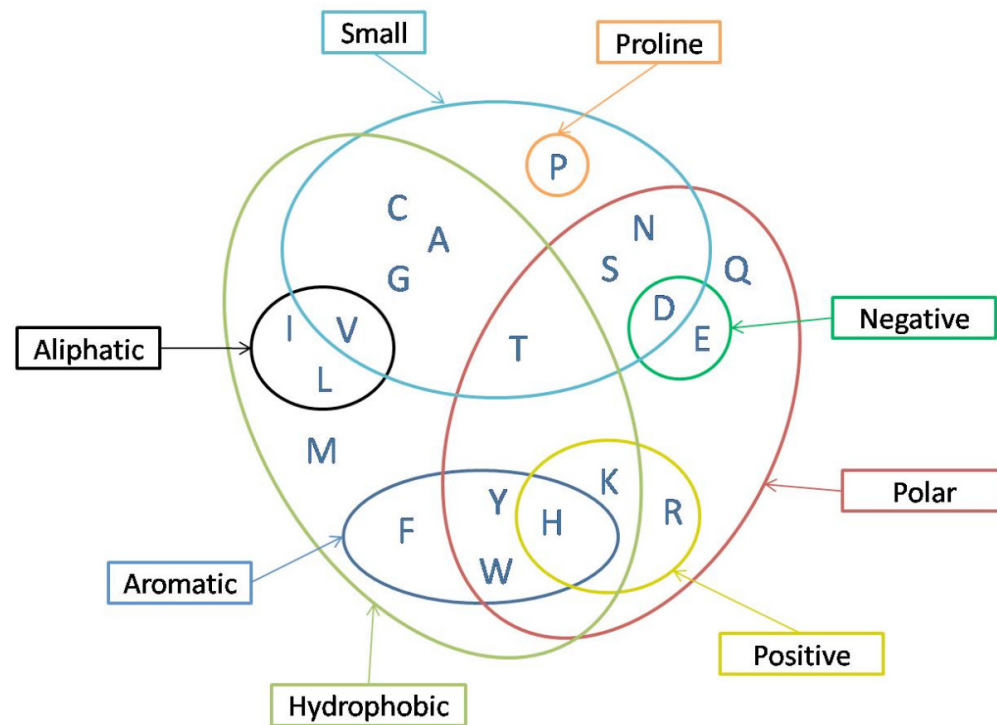
Some more linkers:



The ease of cleavage from the resin depends on the linker used and the stability of the final carbocation:



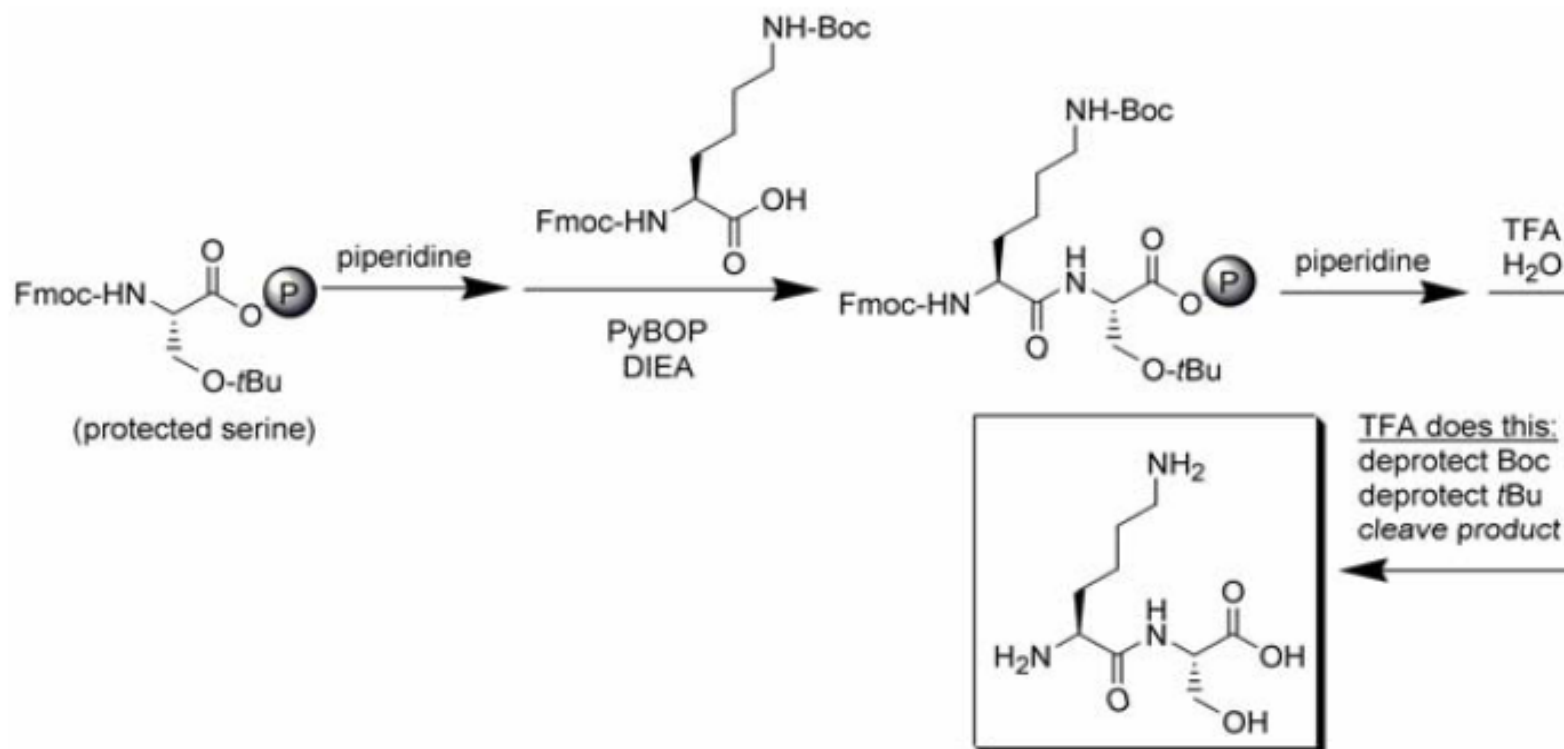
Properties of amino acids



Protecting groups (PG)

How does we deal with reactive side chains of amino acids?

- They are masked/protected with “semi-permanent” PG until cleavage
- Example: H-Lys-Ser-OH



Protecting groups

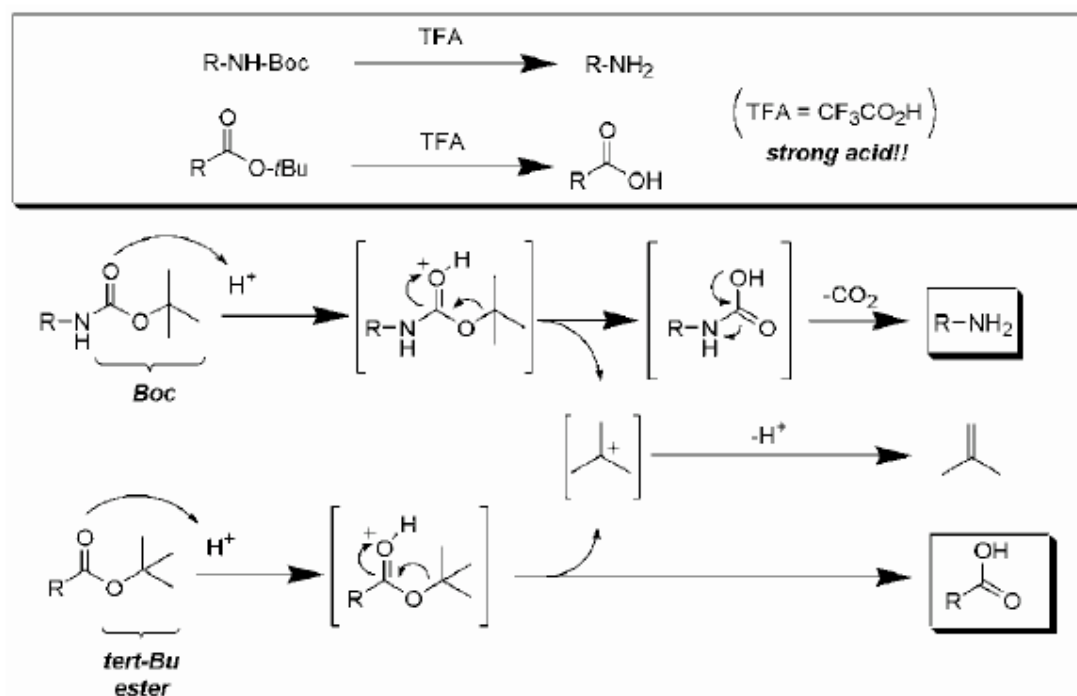
What is a protecting group?

- A fragment bound to a functional group that blocks the reactivity of that group
- Good protecting groups are easily attached and removed using mild reaction conditions



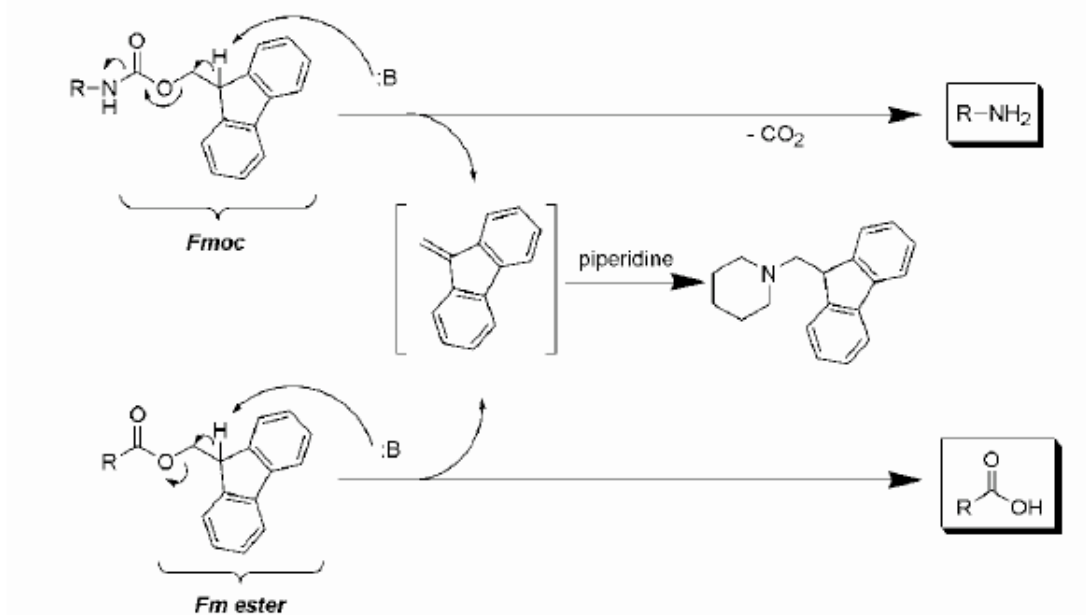
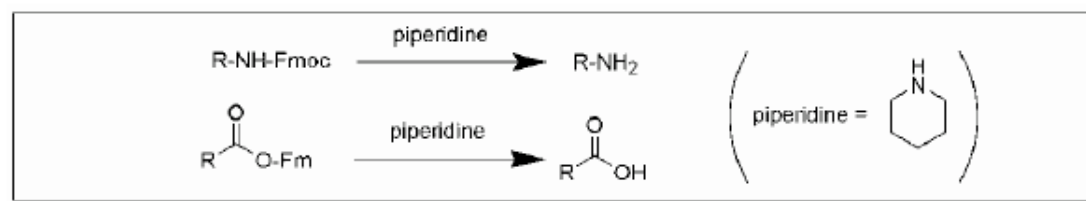
Protecting groups

Acid labile protecting groups: Boc and tert-Bu ester



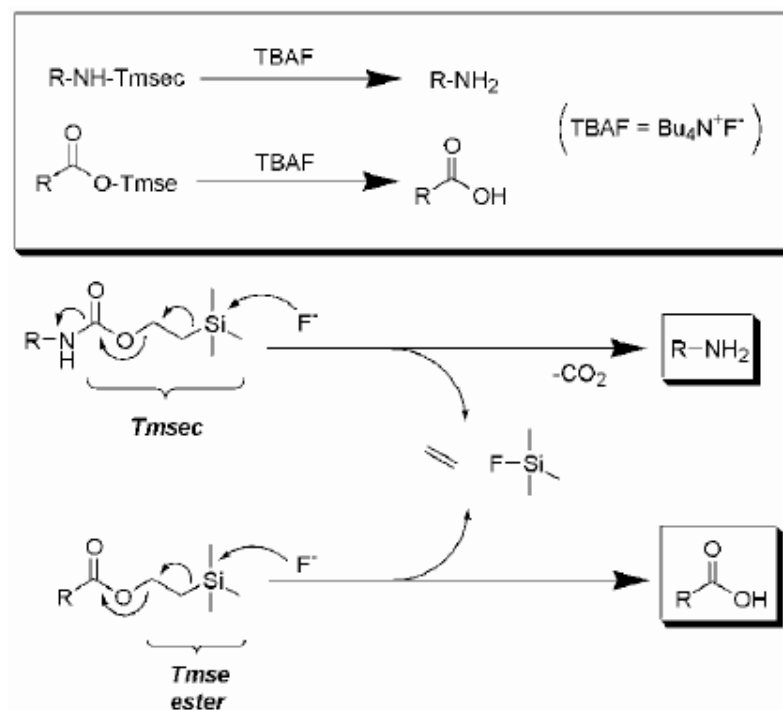
Protecting groups

Base labile protecting groups: Fmoc & Fm ester



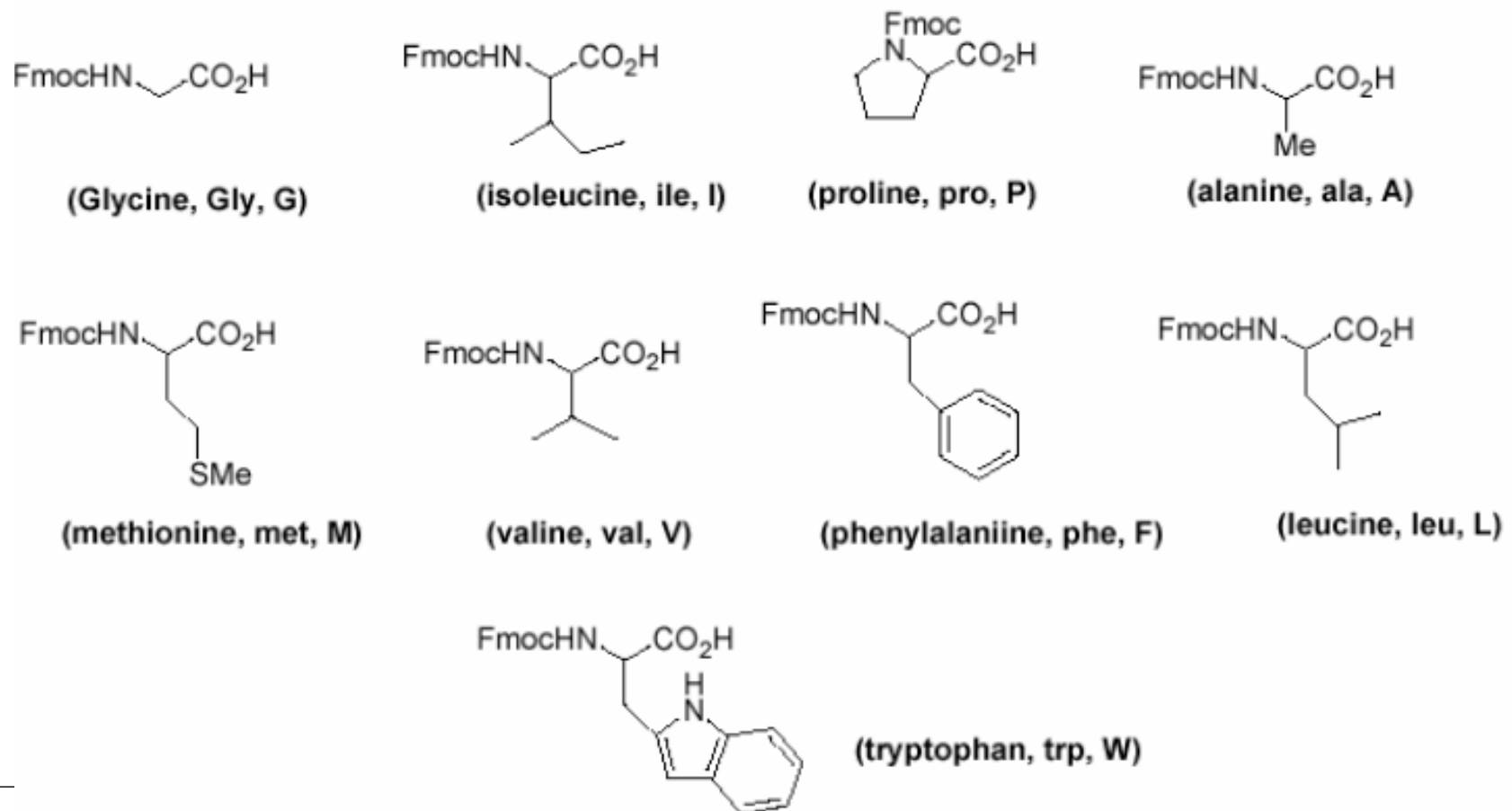
Protecting groups

Fluoride labile protecting groups: Tmsec & Tmse ester



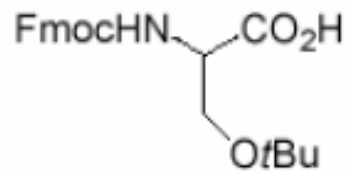
Amino acids

1. side-chain = H, alkyl, aryl –no side-chain protecting groups required

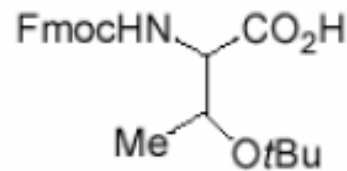


Amino acids

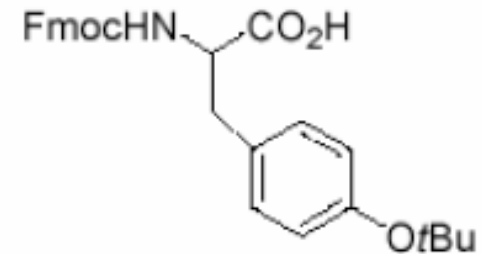
2. side-chain contains OH, protect as *t*Bu ether



(serine, ser, S)

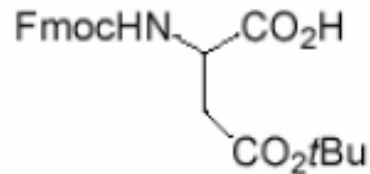


(threonine, thr, T)

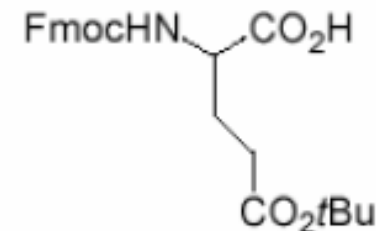


(tyrosine, tyr, Y)

3. side-chain contains CO₂H protected as *t*Bu-ester.



(aspartic acid, asp, D)

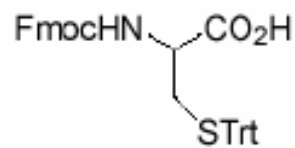


(glutamic acid, glu, E)

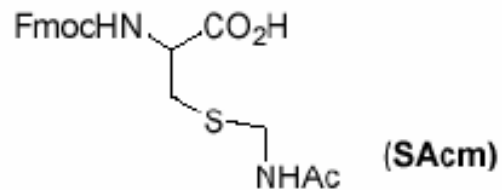


Amino acids

4. side-chain contains thiol (SH) group – depends on application.

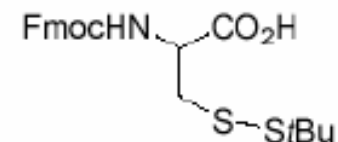


(cysteine, cys, C)



(SAcm)

(acid stable, cleaved with Hg^{2+})

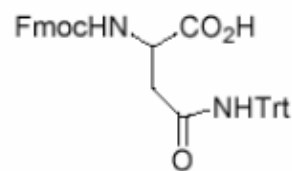


(acid stable, cleaved with thiols)

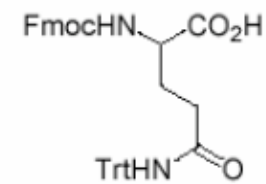


Amino acids

5. side-chain contains CONH₂ group – protect with trityl group.

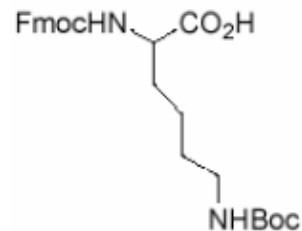


(asparagine, asn, N)

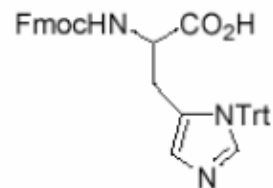


(glutamine, gln, Q)

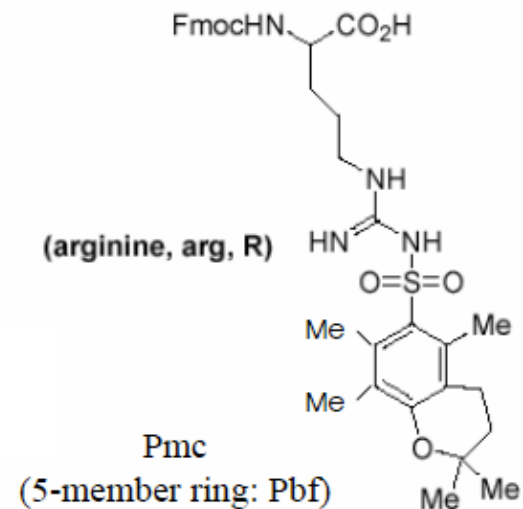
6. side-chains contain basic or imidazole groups.



(lysine, lys, K)



(histidine, his, H)

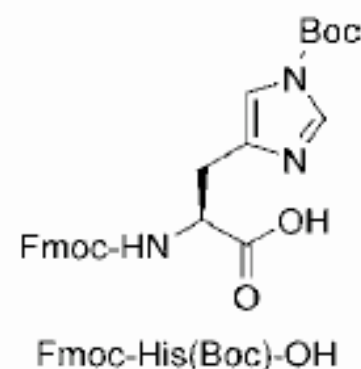
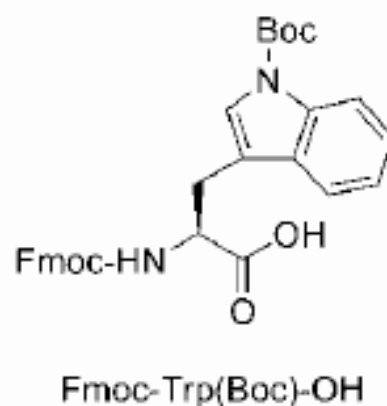
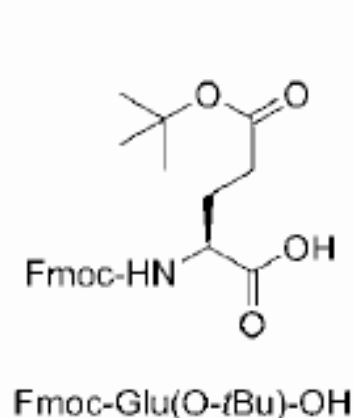
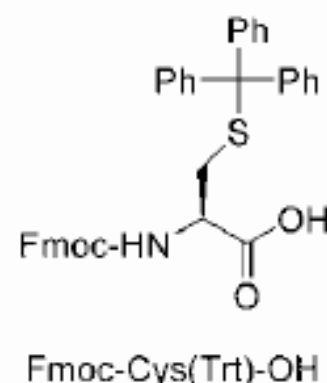
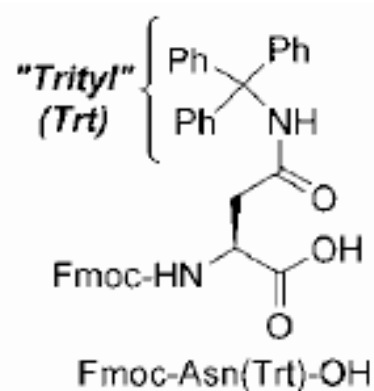
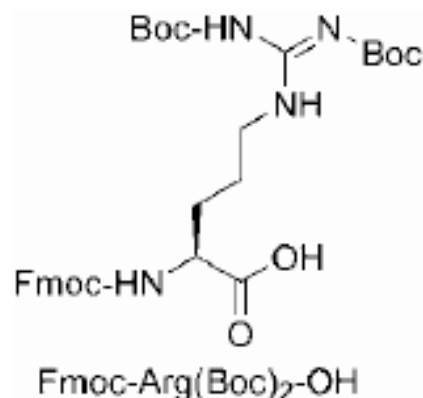


(arginine, arg, R)
Pmc
(5-member ring: Pbf)



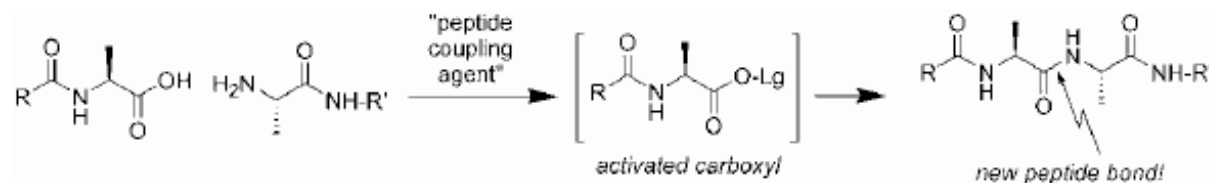
Solid phase synthesis: to make peptides

- Examples of protected AAs with reactive side-chains:
 - Protected side chains cleaved during resin cleavage with TFA!

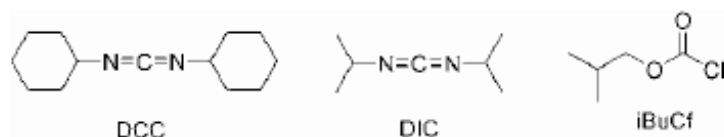


Solid phase synthesis: to make peptides

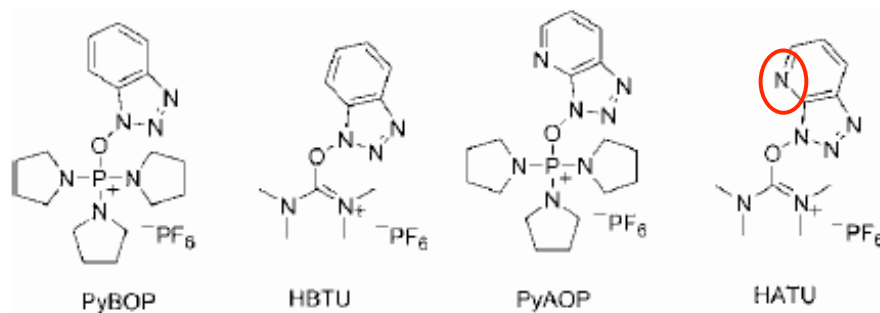
- Peptide bonds created with peptide coupling agents:



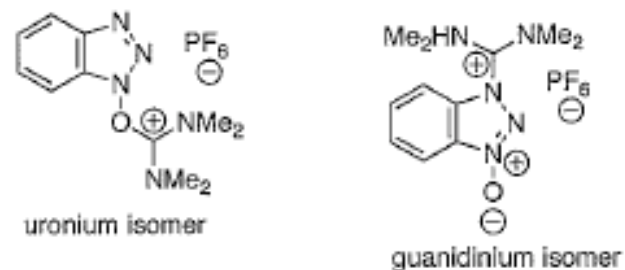
- First generation coupling agents:



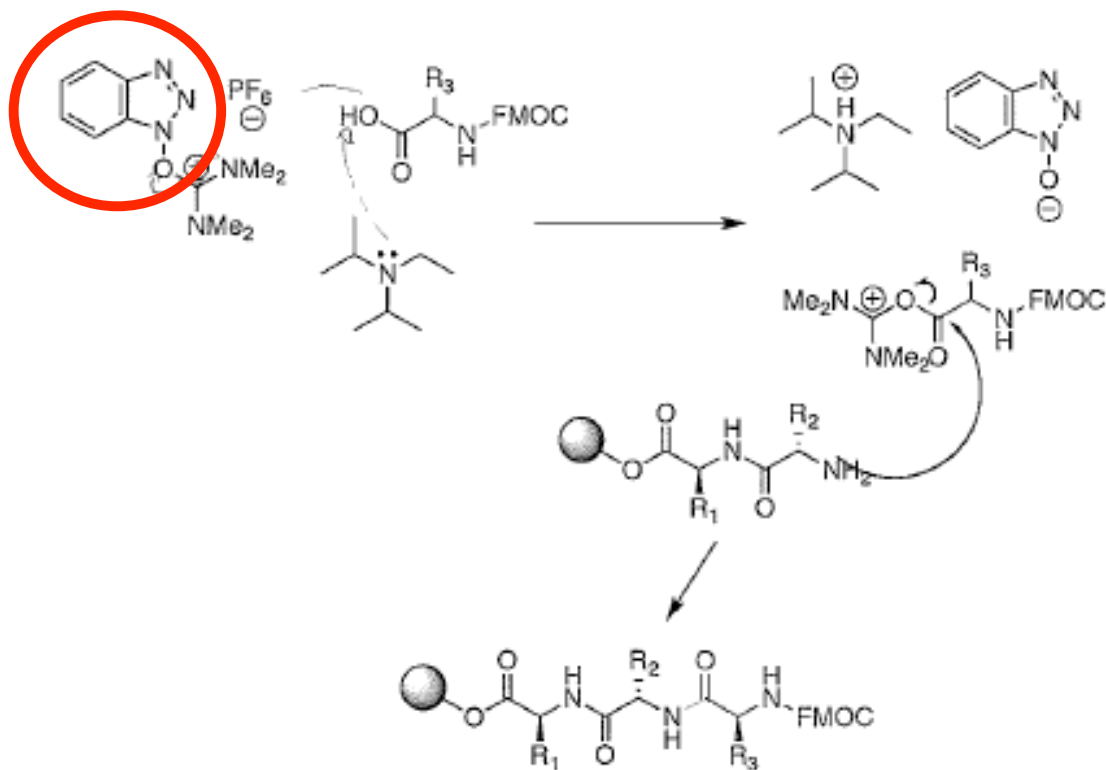
- Later generation coupling agents: (improved in many ways...)



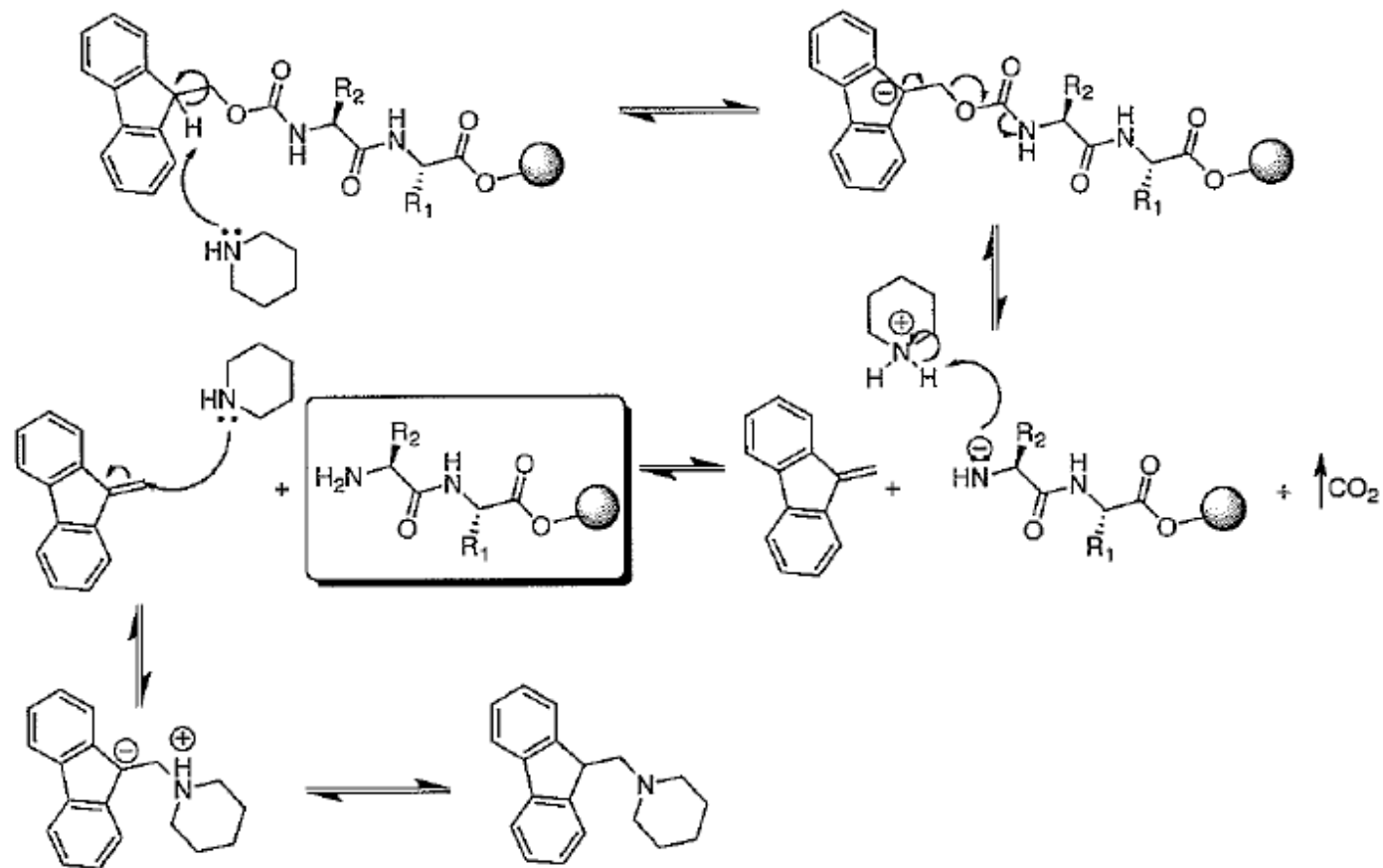
Coupling agents – Activation of amino acids



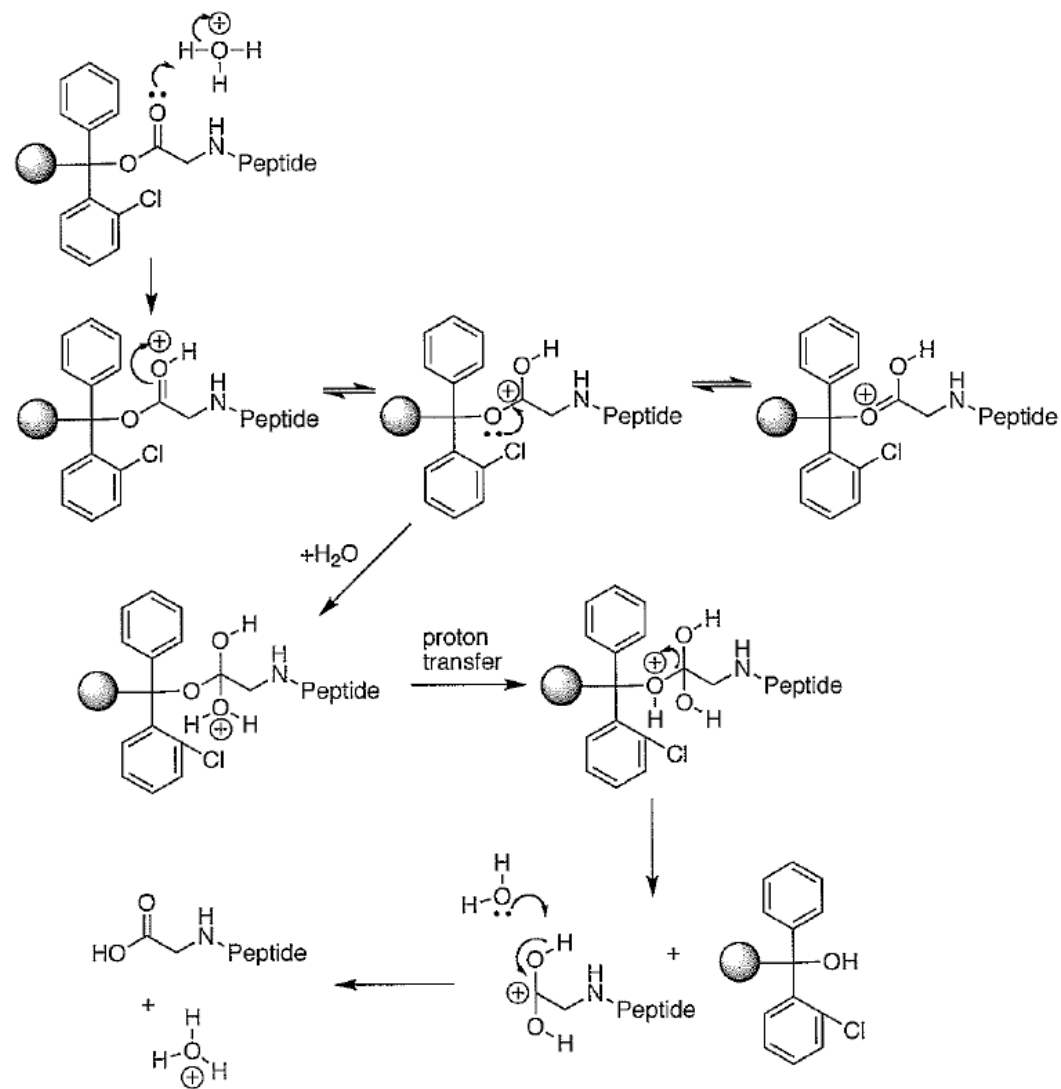
Active ester - good
leaving group



Deprotection of Fmoc by using piperidine



Cleavage reaction using TFA



Cleavage reaction

Mild cleavage – PGs are not removed

Scavengers:

- EDT (ethanedithiol) – scavenger for t-butyl cation, help to remove Trt from Cys
- EDT and thioanisole – avoid Met oxidation
- Phenol – protect Tyr and Trp
- TIS (Triisopropylsilane) – quenching high stable Trt cation

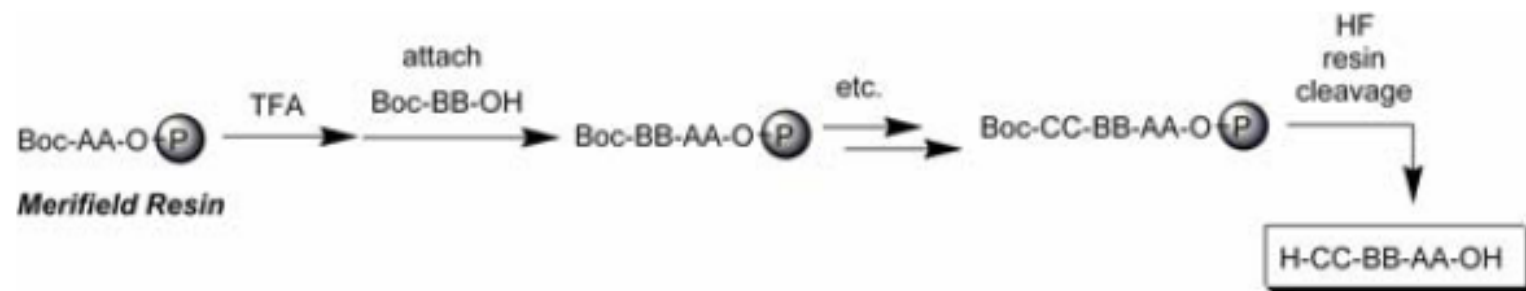
Cleavage

Side reactions during cleavage:

- Alkylation for Met, Cys and Trp (by t-Butyl action)
- Sulfinylation for Trp (by Mtr and Pmc)
 - Use Trp(Boc)

Summary - SPPS

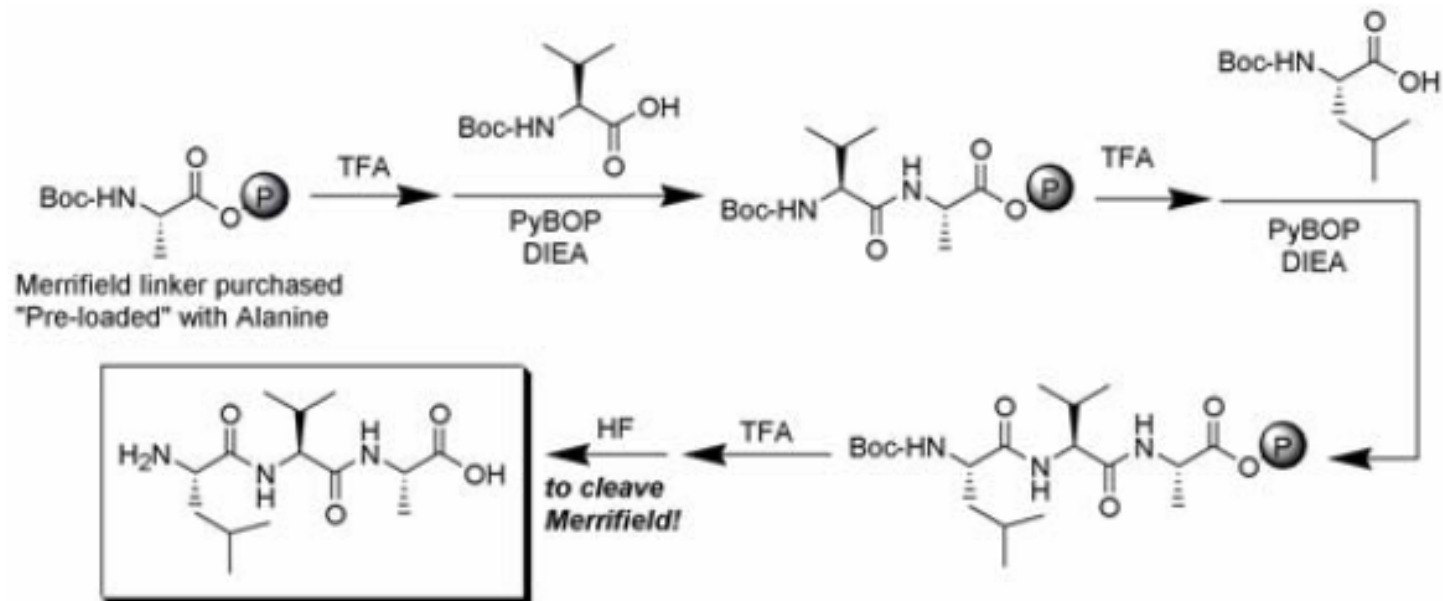
- “Boc-strategy” (Merifield resin)



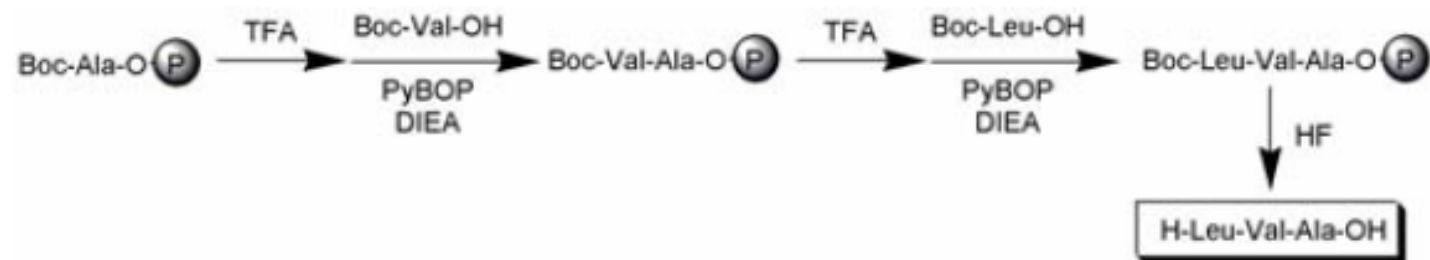
- “Fmoc-strategy” (Wang resin)



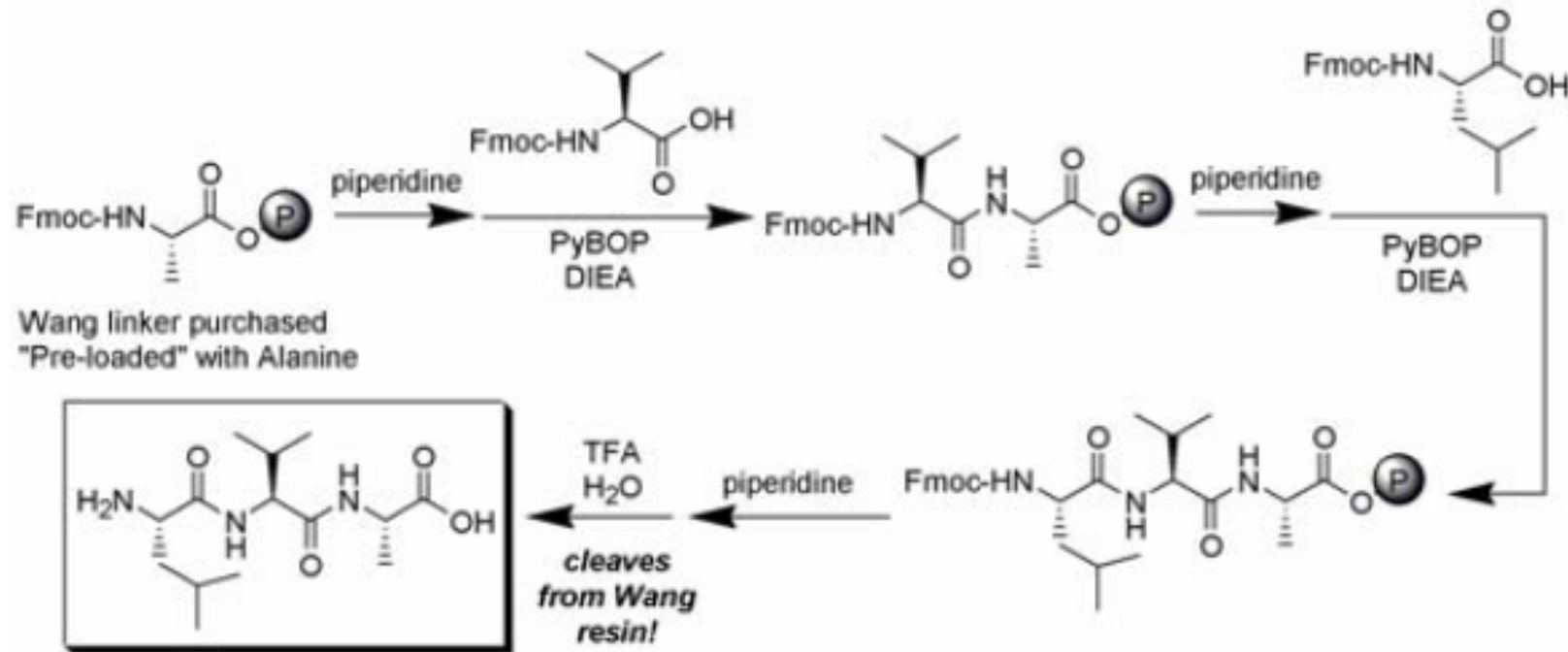
- “Boc strategy” of SPSS: (making H-Leu-Val-Ala-OH):



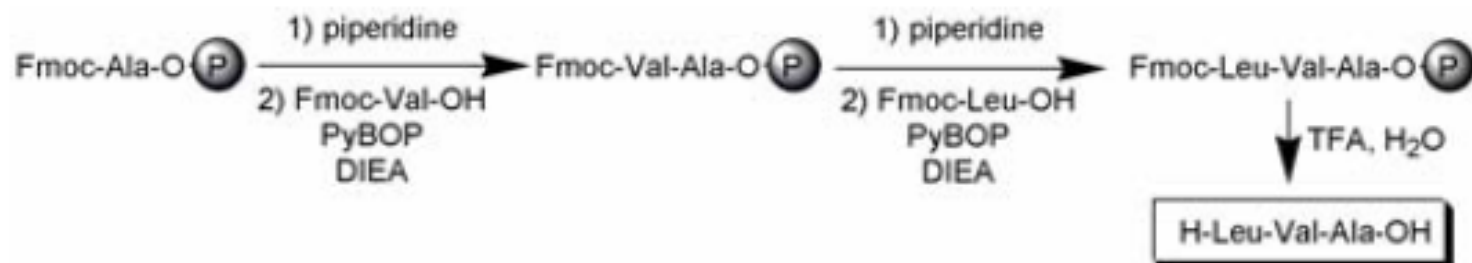
- The sequence can be abbreviated like this:



- And here is the "Fmoc strategy" of SPSS: (H-Leu-Val-Ala-OH):



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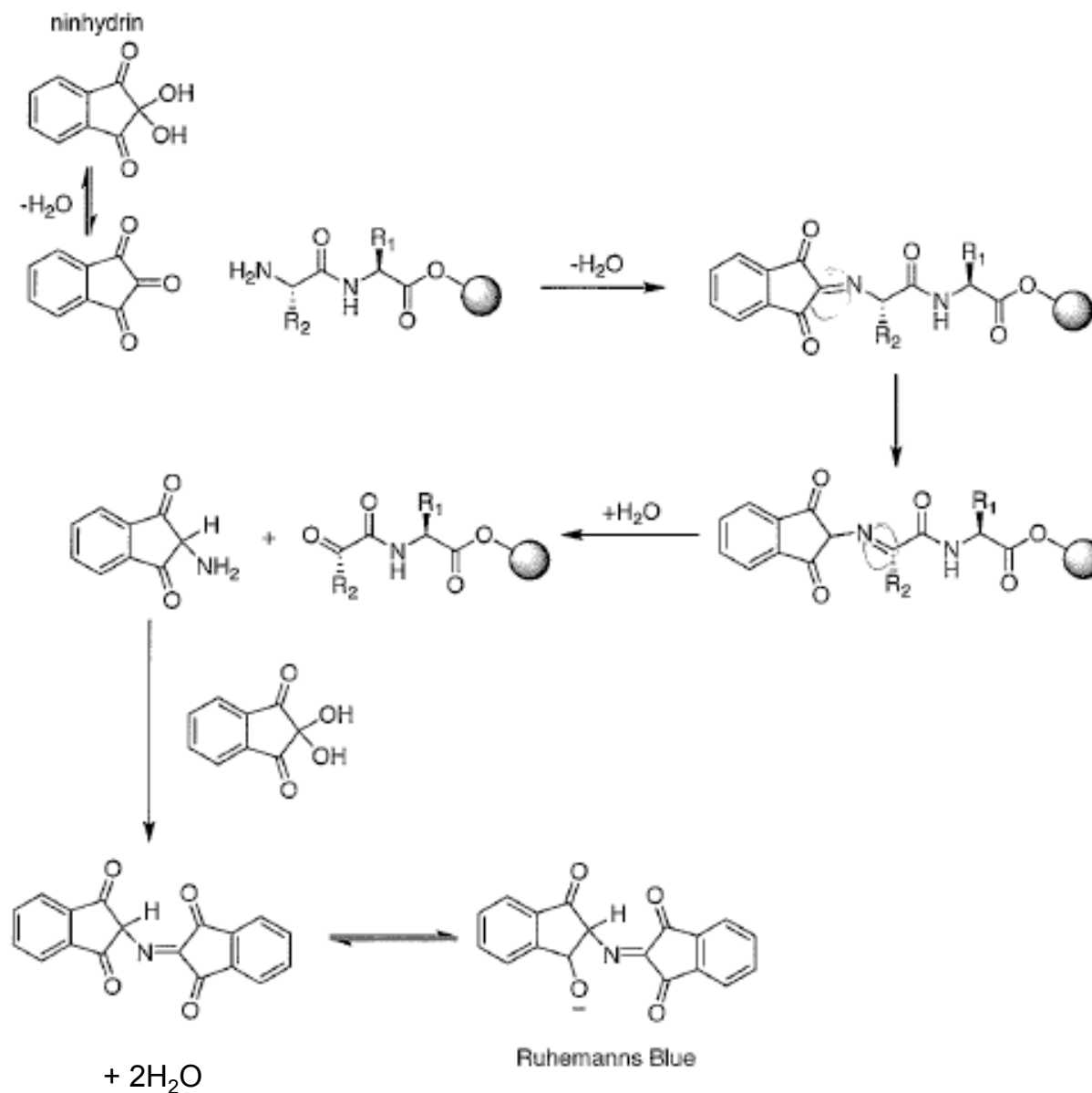
Coupling efficiency and final yield

Efficiency (%)	10-mer	20-mer	30-mer	40-mer	50-mer	60-mer
99	90	82	74	67	61	55
98	82	67	55	45	36	30
95	60	36	21	13	8	5
80	11	1	0	0	0	0



Ninhydrin test

- To detect a free prii
 - Blue colour – fr
 - Yellow colour –



Purification and characterisation

- RP-HPLC
- Mass spectroscopy
- NMR



Leu-Enkephalin

- Pentapeptide
 - NH_2 - Tyr - Gly - Gly - Phe - Leu – COOH
- Using Fmoc based SPPS
- Characterise the peptide by RP-HPLC, NMR and maybe SELDI-MS
- Read “Guide to SPPS” before 1:00 pm
- Check MSDS for all chemicals needed for SPPS before 1:00 pm
 - DMF/Pregnancy
- Two groups
- We’ll meet in N120
- Report

